Acta Crystallographica Section D Biological Crystallography ISSN 0907-4449

Svetlana V. Antonyuk,^a Robert R. Eady,^b Richard W. Strange^a and S. Samar Hasnain^a*

^aMolecular Biophysics Group, CCLRC Daresbury Laboratory, Warrington, Cheshire WA4 4AD, England, and ^bDepartment of Biological Chemistry, John Innes Centre, Norwich NR4 7HU, England

Correspondence e-mail: s.hasnain@dl.ac.uk

The structure of glyceraldehyde 3-phosphate dehydrogenase from *Alcaligenes xylosoxidans* at 1.7 Å resolution

The enzyme glyceraldehyde-3-phosphate dehydrogenase (GAPDH) from the Gram-negative denitrifying bacterial species Alcaligenes xylosoxidans was purified and crystallized as a contaminant protein during purification of nitrous oxide reductase. This is the first structure of a GAPDH from a denitrifying species. The crystal structure was solved at 1.7 Å resolution by molecular replacement using the structure of GAPDH from Bacillus stearothermophilus as a starting model. The quality of the structure enabled the amino-acid sequence of the A. xylosoxidans GAPDH to be assigned. The structure is that of the apo-enzyme, lacking the NAD⁺ cofactor and with the active-site residue Cys154 oxidized. The global structure of the enzyme has a homotetrameric quaternary structure similar to that observed for its bacterial and eukaryotic counterparts. The essential role of Cys154 in the enzyme activity has been confirmed. In monomer O two half-occupancy sulfate ions were found at the active site, which are analogous to the substrate and the 'attacking' phosphate seen in B. stearothermophilus. One half-occupancy sulfate ion is also located in the substrate-binding site of monomer P.

Received 16 September 2002 Accepted 20 February 2003

PDB Reference: glyceraldehyde 3-phosphate dehydrogenase, 10bf, r10bfsf.

1. Introduction

Under anaerobic or micro-aerobic conditions, many aerobic species of bacteria can utilize nitrate instead of oxygen as a terminal electron acceptor. In this process, called denitrification, the stepwise reduction of nitrate to dinitrogen via nitrite, nitric oxide and nitrous oxide is coupled to the synthesis of ATP formed by oxidative phosphorylation. Denitrifiers are widespread and are among the most successful physiological groups of microorganisms in nature. In ecological terms, denitrification is extremely important since it is the route of formation of almost all atmospheric nitrogen and has both agronomic and environmental impact. It results in the loss of soil nitrogen available for plant growth and generation of the greenhouse gas nitrous oxide (N2O). Alcaligenes xylosoxidans is a Gram-negative denitrifying organism that is common in soil and water. The structure and function of the enzyme and ancillary proteins involved in denitrification are the subject of extensive study in our laboratory. During the isolation of nitrous oxide reductase, a contaminating protein present at significant levels at a late stage of purification was resolved and shown by N-terminal amino-acid analysis to be homologous to D-glyceraldehyde-3-phosphate dehydrogenase (GAPDH; EC 1.1.2.12).

© 2003 International Union of Crystallography Printed in Denmark – all rights reserved

GAPDH is a key enzyme in the glycolytic pathway, since in the presence of inorganic phosphate and NAD⁺ it catalyses the reversible oxidative phosphorylation of D-glyceraldehyde3-phosphate (GAP) to 1,3-diphosphoglycerate. The enzyme is composed of four chemically identical subunits, each of which binds one molecule of NAD coenzyme. The binding of the coenzyme shows positive or negative cooperativity, depending on the source of the enzyme.

Crystal structures have been solved for GAPDH enzymes isolated from a number of eukaryotic and bacterial species. Two major interests have driven these studies: firstly to establish the structural basis for the extreme thermostability of GAPDH from the archaea and thermophiles and secondly to gain insight to the basis of the cooperativity of cofactor binding. The range of structural studies is wide and in microbial sources includes the mesophiles Escherichia coli (Duée et al., 1996) and Bacillus coagulans (Griffith et al., 1983), the moderately thermophilic Bacillus stearothermophilus (Skarzynski & Wonacott, 1988; Didierjean et al., 1997), the thermophilic Thermus aquaticus (Tanner et al., 1996) and the hyperthermophilic Thermotoga maritima (Korndörfer et al., 1995). The glycosomal GAPDHs of Trypanosoma brucei (Vellieux et al., 1993), Leishmania mexicana (Kim et al., 1995) and Tryponasoma cruzi (Souza et al., 1998) have been studied. More recently, crystallographic analyses of GAPDHs from Palinurus versicolor (Song et al., 1998), Saccharomyces cerevisiae (Gilboa et al., 1998), spinach chloroplasts (Sabatino et al., 1999; Fermani et al., 2001), archaeal GAPDH from Sulfolobus solfataricus (Fleming et al., 1998; Isupov et al., 1999) and Methanothermus fervidus (Charron et al., 1999) have been reported. For three GAPDH structures, Saccharomyces cerevisiae (Gilboa et al., 1998), Thermoproteus tenax (Brunner et al., 2000) and Synechococcus (Nakamura et al., 2001), only preliminary crystallographic analyses are available. All these studies have revealed similar three-dimensional structures. The highest resolution reported so far is 1.8 Å for holo GAPDH from B. stearothermophilus and E. coli.

Here, we report the crystal structure of GAPDH from *A. xyloxidans* (*Ax*GAPDH) at 1.7 Å resolution, the first for a GAPDH from a denitrifying species, which we show to be essentially in the apo-form lacking the NAD⁺ cofactor. The structure exhibits the classical Rossmann fold and has significant structural homology with known GAPDH structures.

2. Materials and methods

2.1. Purification of GAPDH

Protein was initially purified as a contaminant during the purification of *A. xylosoxidans* nitrous oxide reductase. *A. xylosoxidans* was grown anaerobically at 330 K under denitrifying conditions in a 200 l pilot plant fermenter (New Brunswick Scientific) with minimal stirring at 50 rev min⁻¹ (Abraham *et al.*, 1993). All purification steps were carried out using 10 mM Tris-HCl buffer pH 8.0 (buffer *A*) at room temperature.

Cell paste (500 g) was suspended in 700 ml 50 mM Tris-HCl buffer containing small amount of deoxyribonuclease and then disrupted in a pre-cooled Manton–Gaulin homogeniser

Table 1

Summary of data collection and model refinement statistics.

Data-collection temperature (K)	100
Resolution range† (Å)	50-1.7 (1.76-1.7)
Completeness† (%)	97.6 (90.7)
R_{merge} †‡ (%)	9.9 (35.0)
$I/\sigma(I)$, last shell	2.2
Redundancy	6
Unique reflections	104690
<i>B</i> factor of data from Wilson plot ($Å^2$)	22.7
Unit-cell parameters	a = 89.0, b = 146.6, c = 145.5
Solvent (%)	60.3
Final R_{cryst} (%)	16.8
$R_{\rm free}$ (%)	18.9
ESU (Å)	0.08
G factor	0.04
No. of atoms (proteins)	5572 (4995)
No. of water molecules	560
No. of sulfate ions	3
No. of potassium ions	2
Average B factor ($Å^2$)	22.3
R.m.s. deviation from ideality	
Bond length (Å)	0.012 (0.021)
Planar groups (Å)	0.005 (0.020)
Bond angles (°)	1.497 (1.948)

† Values in parentheses are for the last resolution shell. ‡ $R_{\text{merge}} = \sum_{h} \sum_{J} |\langle I_{h} \rangle - I_{J}(h)| / \sum_{h} \sum_{J} I(h)$, where I(h) is the intensity of reflection h, \sum_{h} is the sum over all reflections and \sum_{J} is the sum over J measurements of the reflection. § $R_{\text{cryst}} = \sum ||F_{o}| - |F_{c}|| / \sum |F_{o}|$. ¶ Values in parentheses are target values.

at 28 MPa. The resulting suspension was then centrifuged at 29 000g for 3 h to remove cell debris and unbroken organisms. The supernatant, consisting of a red-brown clear suspension, was then dialysed overnight against buffer A (2 \times 201). The dialysed solution was loaded onto a DE-52 cellulose column $(17 \times 9 \text{ cm})$ pre-equilibrated with buffer A. After the protein solution had been loaded, the column was eluted with a linear gradient of a 50-200 mM NaCl (3.4 l total volume) in buffer A at a flow rate of 8 ml min⁻¹. GAPDH was eluted as a pinkyellow slightly cloudy band at 90 mM NaCl together with nitrous oxide reductase (N_2OR) . This fraction was then concentrated to a final volume of 30 ml using an Amicon Centriprep 10 (30 kDa molecular-weight cutoff). It was then loaded onto a gel-filtration column (Sephacryl S-200; 53×5 cm) pre-equilibrated with buffer A containing 150 mM NaCl and eluted at a flow rate of 1.5 ml min^{-1} . A purple band (\sim 35 ml) showed N₂OR activity. This fraction was diluted with 3.5 M ammonium sulfate to an ammonium sulfate concentration of 1.2 M and loaded onto a hydrophobic interaction Toyopearl column. Under these conditions, GAPDH present in the sample passed through the column as a single peak, which when analysed by SDS-PAGE gave a single band.

2.2. N-terminal sequence determination

AxGAPDH (200–300 pmol) was loaded onto an SDS/10% polyacrylamide gel with Tris–glycine as running buffer and then electroblotted onto polyvinylidene difluoride membranes (immobilon_P; Millipore Ltd; Walford, Herts, England) using 10 mM CAPS buffer pH 11 as electroblotting buffer. The electroblotted enzyme was then sequenced on an Applied Biosystems Procise Sequencer 490 series.

2.3. Crystallization and data collection

Crystals were grown using the hanging-drop technique. Droplets containing 5 mg ml^{-1} protein, 12% PEG 550, 100 mM potassium thiocyanate, 50 mM MES buffer pH 6.5 were equilibrated over wells containing 24.5% PEG 550, 200 mM potassium thiocyanate, 100 mM MES pH 6.5. Crystals grew in 4–7 d to 1.5 mm in length.

Crystallographic data were collected at 100 K using beamline 9.6 at a wavelength of 0.87 Å, giving 626 101 diffraction intensities to 1.7 Å resolution. After processing the data with HKL2000 (Otwinowski & Minor, 1997), a total of 104 690 unique reflections were obtained with an R_{merge} of 9.9% on

Alcaligenes xylosoxidans	TIRVAINGYGRIGRNILRAHYEGGKSHDIEIVAINDLGDPKTNAHLTRY	49
Alcaligenes eutrophus	MTIKVAINGYGRIGRNVLRAHYEGGKRHDLEIVAINDLGNAATNAHLTQY	50
Shewanella oneidensis	MTIRVAINGYGRIGRNVLRALYESEKAYPIKIVAINDLGDASINAHLTKY	50
Pseudomonas aeruginosa	MTIRLAINGFGRIGRNVLRALYTGHYREQLQVVAINDLGDAAVNAHLFQY	50
Neisseria gonorrhoeae	MGIKVAINGYGRIGRQVLRAIYDYQIQDQLQIVAVNASGSLETNAHLTKF	50
Neisseria meningitidis	MGIKVAINGYGRIGRQVLRAIYDYQIQDQLQIVAVNASGSLETNAHLTKF	50
Bradyrhizobium japonicum	MAVRVGINGFGRIGRNVLRAIAESG-RKDIEVVGINDLGPVETNAHLLRF	49
	:::.***:****::*** :::*.:* **** ::	
Alcaligenes xylosoxidans	DTAHGKFPGTVSVNGSYMVVNGDKIRVDANRNPAQLPWGALKVDVVLECT	99
Alcaligenes eutrophus	DTVHGRFPGEVSVDGDAFRVNGDRIRVLAQRNPAELPWGELGVDVVMECT	100
Shewanella oneidensis	DSVHGRFNAKVEHDAEAIYVNGDKILTFQERDPGKLPWNDLQVDVVFECT	100
Pseudomonas aeruginosa	DSVHGHFPGEVEHDAESLRVMGDRIAVSAIRNPAELPWKSLGVDIVLECT	100
Neisseria gonorrhoeae	DTVHGRFEADVSHDGGNLIVNGDKIPFFSTRNPAELPWKELGVDLVMECT	100
Neisseria meningitidis	DTVHGRFEADVSHDGGNLIVNGDKIPFFSTRNPAELPWKELGVDLVMECT	100
Bradyrnizobium japonicum	DSVHGRFPGTVTVDGDSISLGGGKIKVTAERDPSKLPWKDLGVDIALECT	99
Alaslianna wylagowidana		140
Alcaligenes autrophus	GFF11KEKAGARIKGGAKKV11SAPGGADVDATVV1GVNHG1LKS1D1V1	150
Shewanella oneidensis	GLF ISKERASARLINGGARNVIISAFGGRDVDAIIVIGVINGVLKAIDIVI	140
Breudomonag aeruginoga	CLETCEDKAAAU OACACKULICAEC_KOVEATVVIGVNUEVLEACUETV	149
Neisseria goporrhoeae	GAFTSKEKAKIHI.FSGAKKVLISAPGCDDVDATVVVGVNDSVLTADMTVV	150
Neisseria meningitidis	GAFTSKEKAKTHLESGAKKVLISAPGGDDVDATVV/GVNDSVLTADMTVV	150
Bradyrhizobium japonicum	GIFTSKDKASAHLTAGAKRVLVSAPA-DGADATIVYGVNHDTLTKDHLVV	148
braay millooram japomeam	*:**::: *: .** :*::***:**:*****	110
Alcaligenes xvlosoxidans	SNASCTTNCLAPLVKPLNDKLGLODGLMTTVHAYTNNOVLTDVYHEDLRR	199
Alcaligenes eutrophus	SNASCTTNCLAPLVKPLHEKLGVVNGLMTTVHSYTNDOVLTDVYHEDLRR	200
Shewanella oneidensis	SNASCTTNCLAPFAKPLNDEIGIESGLMTTIHAYTNDORLSDVYHTDLRR	199
Pseudomonas aeruginosa	SNASCTTNCLAPVAQVLHRELGIEHGLMTTIHAYTNDQNLSDVYHPDLYR	199
Neisseria gonorrhoeae	SNASCTINCLSPVAKVLSESVGIVKGAMTTIHALINDQTVTDVRHKDLRR	200
Neisseria meningitidis	SNASCTTNCLSPVAKVLSESVGIVKGAMTTIHALTNDQTVTDVRHKDLRR	200
Bradyrhizobium japonicum	SNGSCTTNCLAPVAKVLNDLVGIETGFMTTIHAYTGDQPTLDTMHKDLYR	198
	.**********************************	
Alcaligenes xylosoxidans	ARSATMSMIPTKTGAAAAVGLVLPELDGKLNGYAIRVPTINVSIVDLSFV	249
Alcaligenes eutrophus	ARSATMSMIPTKTGAAAAVGLVMPELDGRLDGFAVRVPTINVSLVDLSFV	250
Shewanella oneidensis	ARAAALSMIPTKTGAAAAVGLVVPELQGKFDGLAVRVPTVNVSLVDLSFI	249
Pseudomonas aeruginosa	ARSATQSMIPTKTGAAEAVGLVLPELAGKLTGLAVRVPVINVSLVDLTVQ	249
Neisseria gonorrhoeae	ARSGVENMIPTKTGAAKAVGLVLPELKGRLDGLAIRVPTVNVSLVDLSFQ	250
Neisseria meningitidis	ARSGVENMIPTKTGAAKAVGLVLPELKGRLDGLAIRVPTVNVSLVDLSFQ	250
Bradyrhizobium japonicum	GRAAAMSMIPTSTGAAKAIGLVLPELKGKLDGVAIRVPTPNVSVVDLKII	248
	.* ! ****.**** * ! *** *!! * * ! ***. ***!***	
Alcaligenes xylosoxidans	AKKNTTVEEVNGILK-AASEGELKGILDYNTEPLVSVDYNHDPASSTVDA	298
Alcaligenes eutrophus	AARPITVEEVNGILK-AAAEGELKGILDYNTAPLVSVDFNHNPASSTFDA	299
Snewanella oneidensis	ASRDITVEEVNALLERAASVAPLSEVLAVNKEPLVSIDFNHNAFSSNFDA	299
Neigeonia generrheese	VARDISVDEVINKLIK-EASEGSPVLGINIQPLVSVDFNHDPRSSIFDA	290
Neisseria gonornoeae	AARDIIVEEINALMKAASEAGALKGILGINILPLVSMDFNIIIEASIFDA	200
Bradurhizohium japonicum	AKDATDVKEINAAMK-DASEOOLKGILGVTNADNVSIDENUDDUSSTEUE	297
Bradymizobrum Japonicum	* * * * * * * * * * * * * * * *	631
Alcaligenes xvlosoxidans	SLTKVS-GRLVKVSSWYDNEWGESNRMLDTTVALMSAA	335
Alcaligenes eutrophus	TLTKVN-GTLVKVSAWYDNEWGFSNRMLDTAVALAHAR	336
Shewanella oneidensis	TOTRVN-GRLVKVMAWYDNEWGFSNRMLDNAVALMNAK	336
Pseudomonas aeruginosa	NHTKVS-GRLVKAMAWYDNEWGFSNRMLDSALALAAARD	334
Neisseria gonorrhoeae	TLTKVVDGNMVKVFAWYDNEWGFSCQMLNTARRMFGLEVRP	341
Neisseria meningitidis	TLTKVVDGNMVKVFAWYDNEWGFSCQMLNTARRMFGLEVRPLK	343
Bradyrhizobium japonicum	DQTKVQNGTLVRVMSWYDNEWGFSNRMADTAVAIAKVI	335
	: * :*:. :******** :* :.: :	

Figure 1

Multiple alignment of selected known GAPDH sequences from bacterial sources produced using CLUSTALW (1.7). Residues conserved on alignment of the top 20 sequences are marked *; nonconserved amino-acid residues are marked with points or colons representing similar or different properties, respectively. The AxGAPDH sequence was built in agreement with this alignment using the electron-density map at 1.7 Å resolution. The residues for which ambiguities remain are underlined and coloured pink.

equivalent intensities. Prior to data collection, crystals were transferred for 4 h into a solution containing 25% PEG 400 plus 75% mother liquor. Crystals belong to space group $C222_1$, with unit-cell parameters after cryocooling (100 K) of a = 89.0, b = 146.6, c = 145.5 Å. The asymmetric unit contains two subunits with a molecular mass of 37.2 kDa, giving a solvent content of 59%(v/v).

2.4. Structure solution and refinement

The structure of AxGAPDH was solved by the molecularreplacement method using the program MOLREP (Vagin & Teplyakov, 2000) with the B. stearothermophilus (Bs) holo-

> GAPDH monomer (PDB code 1gd1) as a starting model. The rotation function shows two peaks at 11.5σ and 11.2σ (the next highest peak was 4.5 σ), corresponding to two molecules in the asymmetric unit. Their positioning with the translation function was characterized by a CC of 0.42 and an R factor of 50%. The complete tetramer is generated by the symmetry operation (-x, y, -z + 1/2) of space group C222₁.

> The model was refined without NCS restraints by the maximum-likelihood method as implemented in REFMAC5 (Murshudov et al., 2000) and model building was performed with O (Jones, 1978). The quality of the electrondensity map was good enough to build a proper model using sequence information for GAPDH from other microbial species together with the 20 N-terminal residues of AxGAPDH determined by Edman degradation. Solvent molecules were added using ARP (Lamzin & Wilson, 1993). The water molecules were gradually added to the model and were positioned only when well defined positive peaks were present in both $2F_o - F_c$ and $F_o - F_c$ electron-density maps and when they could form hydrogen bond(s) with either protein atoms or other water molecules. A summary of the crystallographic datacollection statistics and the structure refinement are given in Table 1.

3. Results and discussion

3.1. Quality of the model

The crystal structure was determined to a final resolution of 1.7 Å with an R factor of 16.6%. All data in the resolution range from 50 Å were used without a σ cutoff, while 5.0% of randomly distributed reflections were assigned to calculate an $R_{\rm free}$ of 18.9%. The close proximity of $R_{\rm free}$ and R indicates the correctness and high quality of the model.

The final model in the asymmetric unit consists of two subunits, O and P, and the electron density was sufficiently well defined to allow the positioning of 335 residues in each of the two subunits and 560 water molecules in the asymmetric unit. In addition, the model contains several half-occupancy ligands: three sulfate ions (two in monomer O and one in monomer P), two potassium ions and one tetraethyleneglycol fragment chain of a PEG molecule.

The dispersion precision indicator (Murshudov *et al.*, 2000) gives an r.m.s error in coordinates of 0.08 Å. The overall *G* factor is 0.04 (*PROCHECK*; Laskowski *et al.*, 1993), which is within the range expected at 1.7 Å resolution. Of the non-glycine residues, 99.7% fall in the most favoured regions of the Ramachandran plot (Ramakrishnan & Ramachandran, 1965); only Val241 ($\varphi = 135$, $\psi = 90^{\circ}$) lies in the disallowed region. This structural feature is conserved among other GAPDHs (Kim *et al.*, 1995; Vellieux *et al.*, 1993; Skarzynski *et al.*, 1987). The electron density for this residue is good in both subunits and the *B* factors are low. The root-mean-square deviation (r.m.s.d.) of peptide units from planarity is 0.09 Å. The average *B* factor is 23.4 Å² for all protein atoms and about 30 Å² for each of the sulfate ions, potassium ions and PEG-molecule fragments.

Approximately 23.5% of the amino-acid residues are in the α -helical conformation, 27.7% in β -sheets and 4.5% in 3₁₀-helices. During refinement, each subunit was treated independently without imposing NCS fold-symmetry restraints. The r.m.s.d. between C^{α} positions in the two subunits is 0.29 Å. In the last stages of the refinement, H atoms were placed in positions calculated according to geometry criteria.

3.2. Sequence identification for AxGAPDH

The amino-acid sequences of GAPDHs from 376 (63 bacterial) species have been determined and show sequence identities ranging from 18 to 85%. In the case of AxGAPDH, no sequence information was available; therefore, the N-terminal sequence of the first 20 residues was determined by N-terminal protein sequencing. A BLAST (Altschul *et al.*, 1990) search showed this portion of the sequence to have high homology (up to 89%) with a number of GAPDH sequences



Figure 2

The $2F_o - F_c$ electron-density map at the 3σ level clearly shows Phe instead of Leu as in the starting model.

from microbial species (Fig. 1). The B. stearothermophilus holo-GAPDH monomer was taken as a starting model for building the 'X-ray sequence' of AxGAPDH using the 1.7 Å resolution electron density. During refinement and model rebuilding, it was possible to locate 335 amino-acid residues per subunit. A large number of iterative cycles of model building and refinement were performed with the $2F_o - F_c$ electron-density map calculated using zero side-chain occupancy for the non-conserved residues. The electron-density maps revealed differences in the AxGAPDH sequence in comparison with BsGAPDH. All residues with aromatic side chains were clearly visible in the original map and residues situated in the inner part of the molecule could easily be defined. For example, the density at residue 56 was clearly that corresponding to a Phe, not a Leu as in the case of BsGAPDH (Fig. 2). This residue is also a Phe in all other bacterial sequences (Fig. 1). Alignment of the AxGAPDH sequence derived from the built structure compared with the 20 topscoring GAPDH sequences allowed us to check that conserved residues were positioned and assigned correctly. Where residues could not be unambiguously determined from the electron density, i.e. Asn or Asp, Gln or Glu and Val or Thr, the most frequently assigned amino acid from the 'topscoring' sequences was selected. Identities were also assigned based on their hydrogen-bonding patterns with neighbouring residues and water molecules. There is still ambiguity remaining in residues Asp39, Asn63, Val76, Gln84, Thr141, Asn186 and Asn230 (Fig. 1) situated on the surface of the molecule with high solvent accessibility and high temperature factors.

3.3. Structure description and comparison with GAPDH from *B. stearothermophilus*

Since the three-dimensional structures of prokaryotic, eukaryotic and archaeal GAPDHs are highly conserved, the structure of AxGAPDH will be compared with the *B. stearo-thermophilus* enzyme, one of the two for which the highest resolution structure of 1.8 Å is available (Skarzynski & Wonacott, 1988; Duée *et al.*, 1996).

The AxGAPDH subunit monomer has two folding domains, the nucleotide (NAD⁺) binding domain and the catalytic (GAP-binding) domain. The NAD⁺-binding domain, consisting of residues 1–152 and 314–335, is formed by a sixstranded parallel β -sheet with helices on both sides of the sheet and may be represented by a highly conserved double β – α – β – α – β motif fold (Lesk, 1995). The α/β fold of the catalytic domain (residues 153–313) is more variable, mainly consisting of antiparallel β -sheet. Superposition of the AxGAPDH and BsGAPDH amino-acid sequences and dentification of the secondary-structure elements is shown in Fig. 3(*a*).

The AxGAPDH and BsGAPDH subunits were superimposed using β -sheets from both domains. The least-squares algorithm implemented in the program O was then used to obtain the best fit of the C^{α} atoms to give an r.m.s.d. of 1.02 Å (maximum of 4.3 Å) for the 332 matching atoms of the 335 atoms of the sequence. The catalytic domains of the two enzymes superimpose better than the whole subunits, with an r.m.s.d. of 0.48 Å. The catalytic domains of holo *Bs*GAPDH and *Ax*GAPDH are rotated by about 5° (Skarzynski & Wonacott, 1988). The NAD-binding domains show more

A.xylosoxidans B.stearotherm.	SSSSSS HHHHHHHHHHHHH SSSSSSS HHHHHHHHHH
A.xylosoxidans B.stearotherm.	SSSS SSSS SSSSS GGGGG SSSS HHHHHHHHHH SS VSVNGSYMVVNGDKIRVDANRNPAQLPWGALKVDVVLECTGFFTTKEKAGAHIKGGAKKV 119 VSVNGNNLVVNGKEIIVKAERDPENLAWGEIGVDIVVESTGRFTKREDAAKHLEAGAKKV 116 SSSS SSSSS GGGGGGGGGG SSSSS HHHHHHHHH SS *. :* *. :* *.
A.xylosoxidans B.stearotherm.	SS SS GGGGG SSS HHHHHHHHHHHHHHHHS SSSSSSS IISAPGGADVDATVVYGVNHGTLKS-TDTVISNASCTINCLAPLVKPLNDKLGLQNGLMI 178 IISAPA-KNEDITIVMGVNQDKYDPKAHHVISNASCTINCLAPFAKVLHEQFGIVRGMMI 175 SSS SSS GGGGG SSSS HHHHHHHHHHHHHHHHHHHHHHHSSSSSS :*:**: ::*:**: ::*:**: **: *:*:*
A.xylosoxidans B.stearotherm.	SSSS SSS HHHHHHGGGG SSSSSSSS TVHAYTNNQVLTDVYHEDLRRARSATMSMIPTKTGAAAAVGLVLPELDGKLNGYAIRVPT 238 TVHSYTNDQRILDLPHKDLRRARAAAESIIPTTTGAAKAVALVLPELKGKLNGMAMRVPT 235 SSSS SSSS GGGGGGGGGGGG *:*: **** * *** *****
A.xylosoxidans B.stearotherm.	SSSSSSSS HHHHHHHHHHHHHHH SSSS GGGGGG SSSS INVSIVDLSFVAKRNTTVEEVNGILKAASEGELKGILDYNTEPLVSVDYNHDPASSTVD 297 PNVSVVDLVAELEKEVTVEEVNAALKAAAEGELKGILAYSEEPLVSRDYNGSTVSSTID 294 SSSSSSSSS HHHHHHHHHHHHH SSSS GGGGGG SSSS : *:.** : .:::*: : : : : : : : : : : : :
A.xylosoxidans B.stearotherm.	GGGSSS SSSSSSSS HHHHHHHHHHHHHHHHH ASLTKVS-GRLVKVSSWYDNEWGFSNRMLDTTVALMSAA 335 ALSTMVIDGKMVKVVSWYDNETGYSHRVVDLAAYIASKGL 334 SSSSSS SSSSSSSS HHHHHHHHHHHHHHHH * * * * :::. :***** ::: :
A.xylosoxidans B.stearotherm	- Alcaligenes xylosoxidans - Bacillus stearothermophilus Identities = 186/334 (55%)

Figure 3

(a) The amino-acid sequence of the native AxGAPDH enzyme (top) is shown aligned with the BsGAPDH enzyme sequence (bottom) on the basis of structural superposition. Amino-acid sequence and numbering of residues for BsGAPDH are from the X-ray structure (PDB code 1gd1). α -Helices, β -strands and 3_{10} -helices are indicated by the labels H, S and G, respectively. Conserved residues are shown in red. (b) Stereo plot of C^{α} traces of one subunit of AxGAPDH and BsGAPDH superimposed using the program O (Jones, 1978). The subunit C^{α} trace of BsGAPDH is shown in blue.

structural difference than the catalytic domains and can be superimposed with an r.m.s.d. of 1.09 Å for matching C^{α} atoms (Fig. 3b). No additional structural elements were found in AxGAPDH, but there are three insertion residues, Gly23,

> Gly24 and Lys25, which elongate the first α helix. As in the other known GAPDHs, there are no salt bridges between the catalytic and NAD-binding domains.

3.4. Tetrameric structure

The dimer OP generates the tetramer **OPRO** using a crystallographic twofold axis coincident with the molecular axis P. The relative positions of subunits in the AxGAPDH tetramer are similar to those in the BsGAPDH tetramer. The AxGAPDH tetramer is $75 \times 85 \times 91$ Å in size, with solvent-accessible surface area of a 38564 Å^2 , calculated according to the method proposed by Lee & Richards (1971). As in all dehydrogenases, a large part of the monomer surface ($\sim 15\ 000\ \text{\AA}^2$) is hidden deep within the tetramer because of the S-loop region at the core of the structure. The S-loop of subunit O in AxGAPDH interacts with units P and Q, while in BsGAPDH it interacts with subunits R and P. The interface between the Pand O subunits is the largest (2245 $Å^2$) in the structure. 26 strong interactions are formed across this interface, including six salt bridges (Table 2). This interface also contains a large hydrophobic cluster, where five residues (Val180, Pro209, Ile234, Val236, Ile243) of one subunit interact with their counterparts in the other subunit. The O-R interface area is 1905 Å² and includes 19 hydrogen bonds. The O-Q interface area is the smallest (905 $Å^2$) and involves only five hydrogen bonds and one salt bridge. Despite the differences in crystalline environment, the two subunits are structurally very similar and are related by twofold molecular symmetry. The subunits were superimposed using all C^{α} atoms, which gives an r.m.s.d. of 0.30 Å.

3.5. Active site

The electron density provides no evidence of co-enzyme at the position of the NAD⁺-binding site, showing the structure to be essentially that of apo-enzyme. The electron density in the active site clearly reveals the presence of two sulfate ions (average *B* factors are 30 Å² with occupancy 0.5, similar to those seen in the

research papers

enzyme from *B. stearothermophilus*). There were no sulfate ions present in the crystallization media, so it appears that *Ax*GAPDH binds the sulfate ions during the purification process. The first sulfate occupying the position of the C3 phosphate of the substrate (site P_s ; Moras *et al.*, 1975) is located close to the S-loop and makes hydrogen bonds with



the side-chain O atom of Thr184, the guanidinium groups of Arg199 and Arg235 and with two molecules of water. The orientation of the sulfate O atoms is clearly defined in the electron density. The guanidinium groups of Arg199 and Arg235 are equivalent to the guanidinium groups of Arg195 and Arg231 in *Bs*GAPDH (Fig. 4*a*).

The second sulfate ion in the inorganic site (P_i) is bound as Sul339, as seen in the *Bs*GAPDH structure. It forms hydrogen bonds to side-chain atoms of the conserved residues Ser153 and Thr212 and the main-chain N atom of Gly213 and also to four surrounding water molecules. Two of these waters in turn form hydrogen bonds to the side chains of the neighbouring conserved residues Ser155 and His181 (Fig. 4*b*).

The conserved residues Cys154 and His181, which are directly involved in catalysis, are both well defined. There is additional electron density associated with the S atom of Cys153. Analysis of this density indicated that the side chain of Cys154 in both subunits has undergone full oxidation and three fully occupied O atoms are bound to the S atom (Fig. 4c). It is highly probable that oxidation occurred during purification of the protein, since as it was purified as a contaminant and no preventive measures were taken to avoid oxidation of the active cysteine residue. Nevertheless, comparison with other GAPDH structures shows that oxidation does not affect the position of the S atom or other neighbouring protein residues.

His181 ND1 is hydrogen bonded to the carboxyl O atom of Ala182, fixing the plane of the histidine ring, while the other N atom acts as proton acceptor during catalysis. In the present structure, this nitrogen is hydrogen bonded to one of Cys154 O atoms. The distance from the S atom of Cys154 to this nitrogen is 3.9 Å. This region of the active site is arranged very similarly in all other GAPDHs, suggesting the importance of the configuration and supporting the view that a lock-and-key mechanism is operative.



Figure 4

 $2F_o - F_c$ electron-density map around the active site and sulfate ions (contoured at 1σ). (a) P_s binding site; (b) P_i binding site; (c) overview of the active site. Important hydrogen bonds are shown as red dotted lines (lengths in Å).



Figure 5

 $2F_o - F_c$ electron-density map contoured at 1σ around the PEG molecule and potassium ions near the S-loop area at the dimer–dimer interface.

3.6. PEG-binding site

The AxGAPDH structure differs from all other GAPDHs in the S-loop region. It is probable that this difference is primarily a consequence of the binding of a tetraethyleneglycol fragment of a PEG molecule in the axis region of the tetramer core on the twofold axis in the P and Qmolecules. The PEG fragment has an S shape and surrounds two metal ions (probably potassium ions), coordinating to carboxyl O atoms of S-loop residues Asn185, Asn186 and Gln187. Although the atoms of the PEG molecule have an occupancy of 0.5, they have very clearly defined electron density, with the O atoms coordinating to the potassium ions from one side and the carboxyl oxygen of Tyr193 and a water molecule forming hydrogen bonds from the other side (Fig. 5). The electron density of PEG is better defined in the P subunit. It was difficult to model ethyleneglycol fragments in molecule O. Electron density appearing to be similar to two ethyleneglycol fragments was modelled as chain of waters with halfoccupancy.

4. Concluding remarks

The tetrameric structure of the first GAPDH isolated from a denitrifying bacterium has a molecular architecture that closely resembles those determined from a variety of sources. The structure reported here represents the highest resolution for any known GAPDH. The tetrameric structure of AxGAPDH is formed from two dimers and involves eight hydrogen bonds. The catalytic domain in AxGAPDH is rotated by about 1° relative to apo BsGAPDH and 5° relative to holo BsGAPDH. Similar differences are known between apo and holo forms of other GAPDHs. Two sulfate ions are located in the monomer O, one in the position of the substrate phosphate group (GAP) and the other in the inorganic phosphate-ion site. The position of these sulfate ions is in agreement with those in the apo and holo structures of

Table 2

Selected salt-bridge (s.b.) and hydrogen-bond-type interactions across the O-P, O-R and O-Q interfaces.

(a) O-P interface.

Monomer O	Monomer P	Distance (Å)
Asn174 ND2	Ser304 OGx	3.01
Thr178 OG1	Asp245 OD2	2.68
Thr178 OG1	Lys309 NZ	3.14
Arg198 NH1	Leu282 O	2.8
Arg198 NH1	Asp297 OD2	2.9 (s.b.)
Arg198 NH2	Asp297 OD1	3.24
Arg201 NH1	Asp286 OD1	2.80 (s.b.)
Arg201 NH2	Asp286 OD2	2.83 (s.b.)
Ser206 OG	Val285 N	3.22
Met207 N	Ser284 OG	2.91
Tyr232 OH	Asn245 OD2	2.94
Ser304 OG	Asn174 ND2	315
Ser304 OG	Ser311 OG	2.76
Asp245 OD2	Tyr232 OH	2.94
Asp245 OD2	Thr178 OG1	2.73
Leu282 O	Arg198 NH1	2.89
Ser284 N	Met207 O	3.20
Ser284 OG	Met207 N	2.92
Val285 N	Ser206 OG	3.13
Asp286 OD1	Arg201 NH1	2.84 (s.b.)
Asp286 OD2	Arg201 NH2	2.88 (s.b.)
Asp297 OD2	Arg198 NH1	2.90 (s.b.)
Ser304 O	Asn174 ND2	3.26
Ser304 OG	Asn174 ND2	2.96
Lys309 NZ	Thr178 OG1	3.19
Ser311 OG	Tyr232 OH	2.99

(b) O-Q interface.

Monomer O	Monomer Q	
Lys55 NZ	Asp286 O	2.70
Lys113 NZ	Asp130 OD1	2.81 (s.b.)
Asp128 O	Lys113 NZ	2.91
Asp130 OD1	Lys113 NZ	3.51
Glu280 OE1	Tyr49 OH	2.80
Asp286 O	Lyz55 NZ	2.86

(c)	O-R	interface.
(c)	0 11	miterrace.

Monomer O	Monomer R	
Arg11 NH1	Asp191 O	2.97
Arg14 NH2	Asp191 O	3.09
Asp39 N	Tyr193 OH	3.07
Tyr49 OH	Gln280 NE2	3.12
Asp50 OD2	Asp191 OD2	3.22
Thr51 N	Asp191 OD2	2.91
	Asp191 OD1	2.74
Tyr183 OH	Thr204 OG1	2.76
Asn185 OD1	Thr190 N	3.22
Asn185 ND2	Thr190 O	2.91
Thr190 N	Asn185 OD1	3.22
Thr190 O	Asn185 ND2	2.91
Asp191 O	Arg11 NH1	2.97
Asp191 O	Arg14 NH2	3.09
Asp191 OD1	Thr51 OG1	2.74
Asp191 OD2	Asp50 OD2	3.22
Asp191 OD2	Thr51 N	2.97
Tyr193 OH	Asp39 N	3.07
Thr204 OG1	Tyr183 OH	2.76

BsGAPDH. The apo form of AxGAPDH has weak ionprotein interactions at both phosphate-binding sites, in agreement with the lower occupancies of sulfate ions in apo *Bs*GAPDH structure compared with its holoenzyme form.

We would like to thank Svetlana Sedelnikova for her help with protein purification. We also acknowledge support from CCLRC and BBSRC.

References

- Abraham, Z. H. L., Lowe, D. J. & Smith, B. E. (1993). *Biochem. J.* **259**, 587–593.
- Altschul, S. F., Gish, W., Miller, W., Myers, E. W. & Lipman, D. J. (1990). J. Mol. Biol. 215, 403–410.
- Brunner, N. A., Lang, D. A., Wilmanns, M. & Hensel, R. (2000). Acta Cryst. D56, 89–91.
- Charron, C., Talfournier, F., Isupov, M. N., Branlant, G., Littlechild, J. A., Vitoux, B. & Aubry, A. (1999). Acta Cryst. D55, 1353–1355.
- Didierjean, C., Rahuel-Clermont, S., Vitoux, B., Dideberg, O., Branlant, G. & Aubry, A. (1997). J. Mol. Biol. 268, 739–759.
- Duée, E., Olivier-Deyris, L., Fanchon, E., Corbier, C., Branlant, G. & Dideberg, O. (1996). J. Mol. Biol. 257, 814–838.
- Fermani, S., Ripamonti, A., Sabatino, P., Zanotti, G., Scagliarini, S., Sparla, F., Trost, P. & Pupillo, P. (2001). J. Mol. Biol. 314, 527–547.
- Fleming, T. M., Jones, C. E., Piper, P. W., Cowan, D. A., Isupov, M. N. & Littlechild, J. A. (1998). Acta Cryst. D54, 671–674.
- Gilboa, R., Bauer, A. J. & Shoham, G. (1998). Acta Cryst. D54, 1467– 1470.
- Griffith, J. B., Lee, B., Murdock, A. L. & Amelunxen, R. E. (1983). J. Mol. Biol. 169, 963–974.
- Isupov, M. N., Fleming, T. M., Dalby, A. R., Crowhurst, G. S., Bourne, P. C. & Littlechild, J. A. (1999). *J. Mol. Biol.* **291**, 651–660.
- Jones, T. A. (1978). J. Appl. Cryst. 11, 268-272.

- Kim, H., Feil, I. K., Verlinde, C. L. M. J., Petra, P. H. & Hol, W. G. J. (1995). *Biochemistry*, 34, 14975–14986.
- Korndörfer, I., Steipe, B., Huber, R., Tomschy, A. & Jaenicke, R. (1995). J. Mol. Biol. 246, 511–521.
- Lamzin, V. S. & Wilson, K. S. (1993). Acta Cryst. D49, 129-147.
- Laskowski, R. A., MacArthur, M. W., Moss, D. S. & Thornton, J. M. (1993). J. Appl. Cryst. 26, 283–291.
- Lee, B. & Richards, F. M. (1971). J. Mol. Biol. 55, 379-400.
- Lesk, A. M. (1995). Curr. Opin. Struct. Biol. 5, 775-783.
- Moras, D., Olsen, K. W., Sabesan, M. N., Buehner, M., Ford, G. C. & Rossmann, M. G. (1975). J. Biol. Chem. 250, 9137–9162.
- Murshudov, G. N., Vagin, A. A. & Dodson, E. J. (2000). Acta Cryst. D53, 240–255.
- Nakamura, Y., Tada, T., Wada, K., Kinoshita, T., Tamoi, M., Shigeoka, S. & Nishimura, K. (2001). Acta Cryst. D57, 879–881.
- Otwinowski, Z. & Minor, W. (1997). Methods Enzymol. 276, 307– 326.
- Ramakrishnan, C. & Ramachandran, G. N. (1965). *Biophys. J.* 5, 909–933.
- Sabatino, P., Fermani, S., Ripamonti, A., Casseta, A., Scagliarini, S. & Trost, P. (1999). Acta Cryst. D55, 566–567.
- Skarzynski, T., Moody, P. C. E. & Wonacott, J. A. (1987). J. Mol. Biol. 193, 171–187.
- Skarzynski, T. & Wonacott, A. J. (1988). J. Mol. Biol. 203, 1097-1118.
- Song, S., Li, J. & Lin, Z. (1998). Acta Cryst. D54, 558-569.
- Souza, D. H. F., Garratt, R. C., Araujo, A. P. U., Guimaraes, B. G., Jesus, W. P. D., Michels, P. A. M., Hannaert, V. & Oliva, G. (1998). *FEBS Lett.* **424**, 131–135.
- Tanner, J. J., Hecht, R. M. & Krause, K. L. (1996). Biochemistry, 35, 2597–2609.
- Vagin, A. & Teplyakov, A. (2000). Acta Cryst. D56, 1622-1624.
- Vellieux, F. M. D., Hajdu, J., Verlinde, C. L. M. J., Groendijk, H., Read, R. J., Greenhough, T. J., Campbell, J. W., Kalk, K. H., Littlechild, J. A., Watson, H. C. & Hol, W. G. J. (1993). *Proc. Natl Acad. Sci. USA*, **90**, 2355–2359.